BASIC CHEMISTRY

CHEMISTRY

- ✓ Medieval Latin "Alchimia"/ Arabic "Alkimya", the chemistry
- ✓ The study of the composition and structure of the changes in matter
- It is often called the central science because knowledge in this field is a prerequisite for all the other physical and biological sciences.

HISTORICAL DEVELOPMENT OF THE ATOMIC STRUCTURE

ATOM

- It is the smallest possible unit into which matter can be divided, while still maintaining its properties
- ✓ Building blocks of matter
- ✓ It has protons (+), neutrons (0), and electons (-)
 - Democritus was a greek philosopher who first introduced the concept of atoms. He believed that matter is comprises of tiny indivisible particles known as atomos.
 - 2. Dalton's Atomic Theory
 - a. All matter is made up of indivisible particles known as atoms.
 - b. All atoms of the same elements are alike.
 - c. Compounds are formed when atoms of different elements combine in certain proportions.
 - d. atoms are only rearranged but neither created nor destroyed in chemical reactions.
 - Law of Definite Proportion a given compound always contains the same elements in exactly the same proportion by mass.
 - H₂0 ratio of two hydrogen atom with one oxygen atom only. If it becomes ratio of two hydrogen atom with two oxygen atom, it is not already water.
 - 4. Law of Multiple Proportions when two elements can combine to form more than one type of compound, the masses of one element that combine with a fixed mass of the other elements are in a ratio of small whole numbers.

CaO + CO₂ → CaCO₃

- Law of Conservation of Mass states that matter can neither be created nor destroyed. Matter is just transformed.
- 6. The discovery of the different sub-atomic particles:
 - J.J. Thompson determined the charge to mass ratio of the electron using a cathode ray tube.

Basic Chemistry Student Guide

J Dewey

Basic Chemistry Student Guide:

Basic chemistry William S. Seese, 1981 Study Guide for Basic Chemistry Karen C. Timberlake, 2010-05-10 Chemistry Richard Post, Chad Snyder, Clifford C. Houk, 2020-09-16 A practical complete and easy to use guide for understanding major chemistry concepts and terms Master the fundamentals of chemistry with this fast and easy guide Chemistry is a fundamental science that touches all other sciences including biology physics electronics environmental studies astronomy and more Thousands of students have successfully used the previous editions of Chemistry Concepts and Problems A Self Teaching Guide to learn chemistry either independently as a refresher or in parallel with a college chemistry course This newly revised edition includes updates and additions to improve your success in learning chemistry This book uses an interactive self teaching method including frequent questions and study problems increasing both the speed of learning and retention Monitor your progress with self tests and master chemistry quickly This revised Third Edition provides a fresh step by step approach to learning that requires no prerequisites lets you work at your own pace and reinforces what you learn ensuring lifelong mastery Master the science of basic chemistry with this innovative self paced study guide Teach yourself chemistry refresh your knowledge in preparation for medical studies or other coursework or enhance your college chemistry course Use self study features including review questions and guizzes to ensure that you re really learning the material Prepare for a career in the sciences medicine or engineering with the core content in this user friendly guide Authored by expert postsecondary educators this unique book gently leads students to deeper levels and concepts with practice critical thinking problem solving and self assessment at every stage Basic Concepts of Chemistry, Study Guide Leo J. Malone, 1989-02-21 This Third Edition revised to provide smoother transitions between topics employs a concise yet informal approach to basic chemistry organized to help students employ basic math skills and problem solving strategies Writing style is straightforward and presentation incorporates many concrete analogies to clarify new concepts Includes Study Guide and Selected Solutions Manual for Basic Chemistry Karen C. many illustrative worked examples Timberlake, 2016-03-29 The Study Guide and Selected Solutions Manual assists students with the text material It contains learning objectives chapter outlines additional problems with self tests and answers and answers to the odd numbered Resources in Education ,1995-07 **Basic Chemistry and Study Guide and Student Solutions** problems in the text Manual and Cd 4. 0, Fourth Edition Steven S. Zumdahl, 2000-05-01 **Catalog of Copyright Entries. Third Series** Wastewater Operator Certification Study Guide John Giorgi, 2011-01-12 Library of Congress. Copyright Office, 1979 Catalog of the Officers and Students of the University in Cambridge Harvard University, 1891 The Student's Guide to the University of Cambridge ,1880 Proceedings of the 4th International Conference on Innovation in Education, Science and Culture, ICIESC 2022, 11 October 2022, Medan, Indonesia Baharuddin, Hesti Fibriasari, Juniastel Rajagukguk, 2022-12-22 We are delighted to present the Proceedings of the 4th International Conference on Innovation in

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