

Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 1

Q1. At the same temperature and pressure, an equal volume of gases contains the same number of ____.

- a.) Molecules
- b.) Electrons
- c.) Protons
- d.) Particles

Q2. The gas law that relates the volume of a gas to the number of molecules of the gas is ____.

- a.) Gay-Lussac's law
- b.) Avogadro's law
- c.) Boyle's law
- d.) Charles's law

Q3. The mass % of carbon in CO_2 is ____.

- a.) 0.034%
- b.) 27.27%
- c.) 3.4%
- d.) 28.7%

Q4. The empirical formula of benzene is ____.

- a.) CH
- b.) C_6H_6
- c.) C_3H_3
- d.) C_2H_2

Q5. What is the mole fraction of the solute in a 1.00 molal aqueous solution?

- a.) 0.00177
- b.) 0.0344
- c.) 0.0177
- d.) 0.1770

Q6. State the number of significant figures in each of the following numbers:

- i.) 207.35

Basic Training In Chemistry

Robert E. Kohler



Basic Training In Chemistry:

Basic Training in Chemistry Steven Hoenig, 2001-01-31 Acknowledgements Preface 1 General Chemistry 2 Inorganic Chemistry 3 Organic Chemistry 4 Instrumental Analysis 5 Units and Measurements 6 Mathematical Concepts Index *Basic Training in Chemistry* Steven Hoenig, 2007-05-08 Basic Training in Chemistry is unique in that it gathers into one source the essential information that is usually widely dispersed This book can be used as a quick reference guide to the different disciplines of Chemistry the areas covered are General Inorganic Organic and Instrumental Analysis Although comprehensive in nature Basic Training in Chemistry is not meant to replace any standard textbook but rather to be a supplement or additional source of information or even a comprehensive review guide Basic Training in Chemistry is a useful addition to any academic or commercial laboratory setting where access to a wide variety of information is needed The book can be an exceptional source of information for the undergraduate or graduate student as well as for the experienced chemist Anyone needing a single source of information covering several different disciplines will find this book to be an excellent addition to their usual references **So! You Want to Study Chemistry What! You Need to Know** Gaines Bradford Jackson, 2012-03

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institution building and entrepreneurship in science shows how a minor medical speciality evolved into a large and powerful academic discipline Drawing extensively on little used archival sources the author analyses in detail how biomedical science became a central part of medical training and practice The book shows how biochemistry was defined as a distinct discipline by the programmatic vision of individual biochemists and of patrons and competitors in related disciplines It shows how discipline builders used research programmes as strategies that they adapted to the opportunities offered by changing educational markets and national medical reform movements in the United States Britain and Germany The author argues that the priorities and styles of various departments and schools of biochemistry reflect systematic social relationships between that discipline and biology chemistry and medicine Science is shaped by its service roles in particular local contexts This is the central theme The author's view of the political economy of modern science will be of interest to historians and social scientists scientific and medical practitioners and anyone interested in the ecology of knowledge in scientific institutions and professions

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