

Studien über phosphorsaure Salze; von E. Erlenmeyer.

Zweite Abhandlung *).

Monoferrophosphat.

Die Uebereinstimmung in der Zusammensetzung vieler Mangano- und Ferroverbindungen liefs vermuthen, dafs sich auch ein dem Monomanganophosphat entsprechendes Ferrophosphat darstellen lassen würde. Da die Ferroverbindungen in saurer Flüssigkeit weit weniger der Oxydation durch den Sauerstoff der Luft ausgesetzt sind, als in neutraler oder alkalischer Flüssigkeit, so durfte man es für möglich halten, durch Auflösen von Eisen in Phosphorsäure und rasches Abdampfen der Lösung auf dem Wasserbad oder Verdunsten derselben im Exsiccator zu dem sauren Ferrophosphat zu gelangen. Ich habe deshalb in Gemeinschaft mit Herrn Otto Heinrich verschiedene Versuche in dieser Richtung angestellt, die ich, obgleich sie nicht zu dem gewünschten Ziele führten, doch nicht unerwähnt lassen darf.

Wenn man fein gepulvertes metallisches Eisen mit soviel 48 procentiger wässeriger Phosphorsäure übergiefst, dafs gerade saures Salz entstehen könnte, so löst sich das Metall unter stürmischer Gasentwicklung auf. Wird die so entstandene grasgrüne Lösung mit Wasser verdünnt, so scheidet sich ein weifser amorpher Niederschlag aus; wird sie direct auf dem Wasserbad erhitzt, so trübt sie sich, wird aber auf Zusatz einiger Tropfen Phosphorsäure wieder klar. Beim Verdampfen dieser klaren Lösung auf dem Wasserbad bildet sich nach einiger Zeit eine blaugraue, Eisenoxyd und Oxydul ent-

*) Vgl. diese Annalen **190**, 189.

Arsenik Und Phosphorsaure Salze

C. F. Rammelsberg



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