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Margaret Crockett



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fraction x is the ratio of moles of one substance in a mixture to the total number of moles of all substances for a mixture of

two substances $c_e a$ and $c_e b$ the mole fractions of each would be written as

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water n_a 54 g 18 g mol 1 3 mol total number of moles $n_a + n_b$ 0 5 3 3 5 mol mole fraction of solute ethyl alcohol $x_b = \frac{n_b}{n_a + n_b}$ 0 5 3 5 0 1429

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40 g of H_2O molar mass of H_2O is 18gm mole and molar mass of CH_3OH is 32gm mole solution moles of CH_3OH 4 5 32 0 14

mole moles of H_2O 40 18 2 2 moles therefore according to the equation

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web nov 6 2023 x_a mol a mol a mol b and x_b mol b mol a mol b note in the given mixture the sum of all the mole fractions present is equal to one $x_a + x_b = 1$ when the mole fractions are multiplied by 100 they give the mole percentage mole fraction is a unitless and dimensionless expression image will be uploaded soon

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web notice that the mole fraction has no units on it and is written as a decimal value do not change it to percent note of caution you could see the term mole percent it is simply the mole fraction multiplied by 100 for example in the problem just below the mole fraction of cinnamic acid is 0.2885 its mole percent would be 28.85

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esta es su magia se puede preparar de mil formas siempre queda rico gusta a todos y además es sano y económico disfruta

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