
Alevel Chemistry Electrochemistry Study Set With 100% Verified Solutions!!

Give one reason why the emf of this cell changes when the electrodes are connected and current flows - ANSWER>>concentration of the ions decrease as they are getting used up and will eventually cause the emf to equal zero

state, if at all, how the emf of this cell will change if the concentration of IO_3^- ions is increased - ANSWER>>emf will increase due equilibrium shifting to the right allowing more electrons to be accepted and more electrons are deposited

Explain why copper reacts with dilute nitric acid but has no reaction with dilute hydrochloric acid - ANSWER>>the emf must be positive for the reaction to be feasible as the Cu has a more negative electrode potentials then the hydrogen electrodes so it will not be feasible therefore Cu will not donate any electrons

predict and explain whether or not you would expect a similar corrosion reaction to occur with copper in the presence of oxygen and water - ANSWER>>the emf would only be slightly positive so won't happen readily

by considering electron transfer state what is meant by the term oxidising agent - ANSWER>>electron acceptor

state the temperature at which the standard electrode potential is measured and give the concentration of $\text{Co}^{2+}(\text{aq})$ that must be measured - ANSWER>>298K

1 moldm⁻³

state the value of the standard electrode potential for the following half-equation.

Give a reason why it has this value

$\text{H}^+ + \text{e}^- \rightarrow \frac{1}{2}\text{H}_2$ - ANSWER>>0.00V

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AN Whitehead



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Chemistry Philip J. Barratt, Michael Charles Cox, 2000 This study guide provides short revision sessions which include extra activities checkpoints and hints in the margin There is also an in depth resources section which explains the skills and techniques needed for exam success **New Understanding Chemistry for Advanced Level Third Edition** Ted Lister, Janet Renshaw, 2000 Matches the specifications of the Awarding Bodies AQA NEAB AEB OCR and Edexcel This accessible text includes frequent hints questions and examination questions providing support and facilitating study at home It features photographs and comprehensive illustrations with 3D chemical structures Cambridge International AS and A Level Chemistry Revision Guide Judith Potter, Peter Cann, 2015-10-29 A revision guide tailored to the AS and A Level Chemistry syllabus 9701 for first examination in 2016 This Revision Guide offers support for students as they prepare for their AS and A Level Chemistry 9701 exams Containing up to date material that matches the syllabus for examination from 2016 and packed full of guidance such as Worked Examples Tips and Progress Check questions throughout to help students to hone their revision and exam technique and avoid common mistakes These features have been specifically designed to help students apply their knowledge in exams Written in a clear and straightforward tone this Revision Guide is perfect for international learners *Longman A-level Study Guide: Chemistry* M. Cox, 1997 Each guide in this series gives expert advice on planning a revision programme detailing how students should pace themselves organize their workload and make the most of their time and energy This edition has been updated to cover the revised syllabus with effect from June 1997 Resources in Education , 1999-04 **Review Notes and Study Guide to Shakespeare** Margaret Loftus Ranald, 1964 Cambridge International AS and A Level Chemistry Revision Guide David Bevan, 2011 Focus your revision build your confidence and strengthen your exam technique with this series of revision guides for AS and A level international sciences **Teaching and Learning STEM** Richard M. Felder, Rebecca Brent, 2024-03-13 The widely used STEM education book updated Teaching and Learning STEM A Practical Guide covers teaching and learning issues unique to teaching in the science technology engineering and math STEM disciplines Secondary and postsecondary instructors in STEM areas need to master specific skills such as teaching problem solving which are not regularly addressed in other teaching and learning books This book fills the gap addressing topics like learning objectives course design choosing a text effective instruction active learning teaching with technology and assessment all from a STEM perspective You ll also gain the knowledge to implement learner centered instruction which has been shown to improve learning outcomes across disciplines For this edition chapters have been updated to reflect recent cognitive science and empirical educational research findings that inform STEM pedagogy You ll also find a new section on actively engaging students in synchronous and asynchronous online courses and content has been substantially revised to reflect recent developments in instructional technology and online course development and delivery Plan and deliver lessons that actively engage students in person or online Assess students

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