

# DRIVE & MEMORY Troubleshooting Pocket Reference



# **Drive And Memory Troubleshooting Pocket Reference**

Robert Bruce Thompson, Barbara Fritchman Thompson

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calorimeter with a total heat capacity of 15 48 kj c the temperature of the calorimeter and its contents changes from 19 77 c to 33 90 c calculate e and h for the following reaction as written 2 nh 3 g 3 o 2 g 2 ch

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web choose 1 answer the final temperature is closer to t 1 than to t 2 a the final temperature is closer to t 1 than to t 2 the final temperature is exactly halfway between t 1 and t 2 b the final temperature is exactly halfway between t 1 and t 2 the final temperature is closer to t 2 than to t 1 c

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web in this set of practice questions we will go over the main types of questions on calorimetry including the heat capacity the heat of reaction finding the final temperature of a mixture constant pressure calorimetry and constant volume calorimetry a few important concepts and formulas you will need to solve these questions

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web 6 a 218 g sample of steam at 121oc is cooled to ice at 14oc find the change in heat content of the system 7 in going from ice at 34oc to steam at 138oc a sample of water absorbs 1 41 x 105 j find the mass of the sample answers 1 2 68 x 104 j 2 6  $23 \times 104$  j 3 1 71 x 104 j 4 1 11 x 105 j 5 2 11 x 105 j 6

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